Solution:

We use the following notation:

- t temperature of the final equilibrium state,
- $t_0 = 0$  °C the melting point of ice under normal pressure conditions,
  - $M_2$  final mass of water,
  - $M_3$  final mass of ice,
- $m'_2 \leq m_2$  mass of water, which freezes to ice,
- $m'_3 \leq m_3$  mass of ice, which melts to water.

a) Generally, four possible processes and corresponding equilibrium states can occur:

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1.  $t_0 < t < t_2$ ,  $m'_2 = 0$ ,  $m'_3 = m_3$ ,  $M_2 = m_2 + m_3$ ,  $M_3 = 0$ . Unknown final temperature t can be determined from the equation

$$(m_1c_1 + m_2c_2)(t_2 - t) = m_3c_3(t_0 - t_3) + m_3l + m_3c_2(t - t_0).$$
(7)

However, only the solution satisfying the condition  $t_0 < t < t_2$  does make physical sense.

2.  $t_3 < t < t_0, m'_2 = m_2, m'_3 = 0, M_2 = 0, M_3 = m_2 + m_3.$ Unknown final temperature t can be determined from the equation

$$m_1c_1(t_2-t) + m_2c_2(t_2-t_0) + m_2l + m_2c_3(t_0-t) = m_3c_3(t-t_3).$$
(8)

However, only the solution satisfying the condition  $t_3 < t < t_0$  does make physical sense.

3.  $t = t_0, m'_2 = 0, 0 \le m'_3 \le m_3, M_2 = m_2 + m'_3, M_3 = m_3 - m'_3.$ Unknown mass  $m'_3$  can be calculated from the equation

$$(m_1c_1 + m_2c_2)(t_2 - t_0) = m_3c_3(t - t_3) + m'_3l.$$
(9)

However, only the solution satisfying the condition  $0 \le m'_3 \le m_3$  does make physical sense.

4.  $t = t_0, 0 \le m'_2 \le m_2, m'_3 = 0, M_2 = m_2 - m'_2, M_3 = m_3 + m'_2.$ Unknown mass  $m'_2$  can be calculated from the equation

$$(m_1c_1 + m_2c_2)(t_2 - t_0) + m'_2 l = m_3c_3(t_0 - t_3).$$
(10)

However, only the solution satisfying the condition  $0 \le m'_2 \le m_2$  does make physical sense.

b) Substituting the particular values of  $m_1$ ,  $m_2$ ,  $m_3$ ,  $t_2$  and  $t_3$  to equations (7), (8) and (9) one obtains solutions not making the physical sense (not satisfying the above conditions for t, respectively  $m'_3$ ). The real physical process under given conditions is given by the equation (10) which yields

$$m_2' = \frac{m_3c_3(t_0 - t_3) - (m_1c_1 + m_2c_2)(t_2 - t_0)}{l}$$

Substituting given numerical values one gets  $m'_2 = 0.11$  kg. Hence, t = 0 °C,  $M_2 = m_2 - m'_2 = 0.89$  kg,  $M_3 = m_3 + m'_2 = 2.11$  kg.